

# TO FIND THE ELECTRONIC AND MAGNETIC PROPERTIES OF DOUBLE PEROVSKITE $\text{La}_2\text{MMnO}_6$ ( $\text{M}=\text{Co}, \text{Ni}$ )



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# Introduction/Motivation

## Single & Double Perovskites

- Perovskite is calcium titanate ( $\text{CaTiO}_3$ ) founded by Gustav Rose named given after Lev Von Perovski.
- Generally single perovskite is  $\text{ABX}_3$  having 'A' as monovalent cation, 'B' as divalent metals cation and 'X' is a halide.
- After the complex replacement of lead in single perovskite a new structure is formed called Double Perovskites  $\text{A}_2\text{B}'\text{B}''\text{X}_6$  where 'A' is a large cation and B', B'' are either trivalent or monovalent cations and 'X' is either Oxygen or Halogens<sup>1</sup>.

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<sup>1</sup>T. Tesfamichael, M. Roknuzzaman, C. Zhang, K. Ostrikov, A. Du, H. Wang, and L. Wang, *Scientific Reports* **9**, 718 (2019).

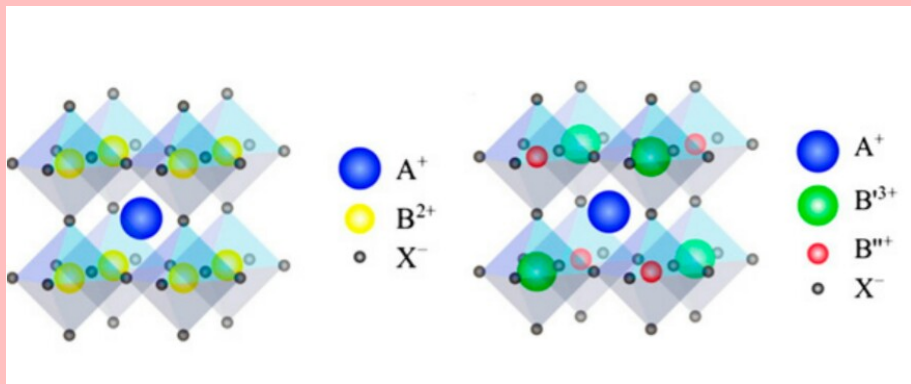


fig1:Single & Double Perovskite<sup>2</sup>

<sup>2</sup>E. L. Meyer, D. Mutukwa, N. Zingwe, and R. Taziwa, *Metals* **8**, 667 (2018).

# Introduction/Motivation

## Applications of Perovskites

- Applications
  - Spintronics devices
  - Light emitting diodes
  - Multistate data storage
  - In photovoltaic research
- So, due to all these applications and properties it encourage me to work in this field.

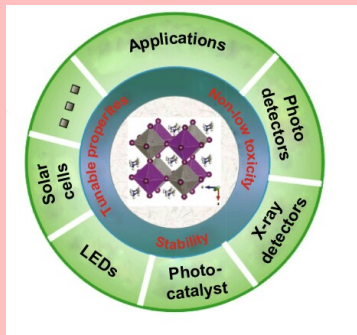


fig2:Application field of Perovskites.<sup>a</sup>

<sup>a</sup>X. Zhao *et.al* *Joule* 2, 1662 (2018).

# Literature Review

- M.P Singh *et al.* (2009) studied  $La_2NiMnO_6$ , found that ordered phase has monoclinic but the disordered phase has pseudocubic structure also both can show the transition from ferromagnetic to paramagnetic<sup>3</sup>.
- A.Kolchinskaya *et al.*(2012) studied  $La_{2-x}Sr_xCoIrO_6$  and found it is antiferromagnetic also magnetic moment depends on the order of the heavy Ir ions<sup>4</sup>.

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<sup>3</sup>M. Singh, K. Truong, S. Jandl, and P. Fournier, *Physical Review*.**79** (2009).

<sup>4</sup>A. Kolchinskaya, P. Komissinskiy, M. Baghaie Yazdi, M. Vafaei Khanjani and D. Mikhailova, *et al. Physical Review. B* **85**, 22 (2012).

# Literature Review

- Z.Y Wu *et al.*(2013) observed  $La_2NiMnO_6$  for the adsorption of bovine serum albumin protein and found that it has highest adsorbtion capacity at  $850^0C$  it adsorb 219.6 mg/g shows that it is very useful in the biomedical<sup>5</sup>.
- G.Kafle *et al.* (2015) studied  $Nd_2MgIrO_6$  and found it is Antiferromagntic in ground state also it is Mott-Hubbard type insulator with space group of  $P2_1/n$  ,Monoclinic distorted double perovskite result potential  $V_{Nd} = 6\text{ev}$  , and  $V_{Ir} = 1.25\text{ ev}$  Also Nd couples antiferromagnetic with Ir<sup>6</sup>.

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<sup>5</sup>Z.-Y. Wu, C.B. Ma, X.G. Tang, R. Li and Q.X. Liu, *et al. Nanoscale Research Letters*. **8**, 207 (2013).

<sup>6</sup>M. Ghimire, G. Kaphle, and R. Thapa, *Journal of Nepal Physical Society* **3**, 50 (2016).

# Literature Review

- P. Kumar *et al.* (2016) observed by doping  $La$  in  $Sr_{(2-x)}La_xNiMoO_6$  and found that electric conductivity is high for concentration 0.04<sup>7</sup>.
- E. Meyer *et al.* (2018) found the Goldschmidt tolerance factor( $t$ ) of halide perovskite were  $0.81 \leq t \leq 1.0$  which also gives the concept of lanthanide based halide double perovskite<sup>8</sup>.

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<sup>7</sup>P. Kumar, N. K. Singh, G. Gupta, and P. Singh, *RSC Advanced*, **6** (2016).

<sup>8</sup>E. L. Meyer, D. Mutukwa, N. Zingwe, and R. Taziwa, *Metals*, **8**, 667 (2018).



# Objectives

- **General Objectives**

- To identify the ground state electronic configurations of  $La_2MMnO_6$  compound.
- To study the structural properties of  $La_2MMnO_6$  compound.
- To compare the obtained result with the available experimental results.

- **Specific Objective**

- Study of the electronic and magnetic properties of  $La_2MMnO_6$  compound.

# Theoretical background and methodology

## Many body Hamiltonian

$$\hat{H}\psi_i(\mathbf{R}_I, \mathbf{r}_i) = E_i\psi_i(\mathbf{R}_I, \mathbf{r}_i) \quad (1)$$

where the hamiltonian can be written as,

$$\hat{H} = \hat{T} + \hat{V} \quad (2)$$

$$\hat{H} = \sum_i \frac{\mathbf{p}_i^2}{2m_i} + \sum_I \frac{\mathbf{P}_I^2}{2M_I} - \sum_I \frac{Z_I e^2}{|\mathbf{r} - \mathbf{R}_I|} + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} + \frac{1}{2} \sum_{I \neq J} \frac{Z_I Z_J e^2}{|\mathbf{R}_I - \mathbf{R}_J|} \quad (3)$$

- K. E. of electrons and K. E. of nuclei.
- P. E due to the attraction between electron-nucleus.
- P. E due to repulsive between electron-electron .
- P. E due to repulsive between nucleus-nucleus.

## Born-Oppenheimer Approximation

$$\hat{H} = \sum_i \frac{\mathbf{p}_i^2}{2m_i} - \sum_I \frac{Z_I e^2}{|\mathbf{r} - \mathbf{R}_I|} + \frac{1}{2} \sum_{i \neq j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} \quad (4)$$

- K. E. of nuclei (neglected) because  $M/m \gg 10^3$  so  $V \ll v$  (molecular confirmation)
- Nucleus-nucleus repulsive P. E (constant).

## Hartree-Fock Approximation:

System of N electrons, the antisymmetrized wave function is given by the following slater determinant.

$$\psi = \frac{1}{\sqrt{N!}} \begin{vmatrix} \phi_1(\mathbf{r}_1, \mathbf{s}_1) & \phi_1(\mathbf{r}_2, \mathbf{s}_2) & \dots & \phi_1(\mathbf{r}_N, \mathbf{s}_N) \\ \phi_2(\mathbf{r}_1, \mathbf{s}_1) & \phi_2(\mathbf{r}_2, \mathbf{s}_2) & \dots & \phi_2(\mathbf{r}_N, \mathbf{s}_N) \\ \dots & \dots & \dots & \dots \\ \phi_N(\mathbf{r}_1, \mathbf{s}_1) & \phi_N(\mathbf{r}_2, \mathbf{s}_2) & \dots & \phi_N(\mathbf{r}_N, \mathbf{s}_N) \end{vmatrix} \quad (5)$$

Hartree-Fock equations is given as follows:

$$\left[ \sum_i \frac{\mathbf{p}_i^2}{2m} - \sum_{i=1}^N \sum_{l=1}^M \frac{Z_l e^2}{|\mathbf{r}_i - \mathbf{R}_l|} + \sum_{j \neq i} \int d\mathbf{r}' \phi_j^*(\mathbf{r}') \frac{e^2}{|\mathbf{r} - \mathbf{r}'|} \phi_j(\mathbf{r}') \right] \phi_i(\mathbf{r}) - \sum_{j \neq i} \left[ \int d\mathbf{r}' \phi_j^*(\mathbf{r}') \frac{e^2}{|\mathbf{r} - \mathbf{r}'|} \phi_i(\mathbf{r}') \delta_{\sigma_i \sigma_j} \right] \phi_j(\mathbf{r}) = \varepsilon_i \phi_i(\mathbf{r}) \quad (6)$$

# Density Functional Theory DFT

- The HF approximation mostly used in pre DFT era was based on Slater determinant form of single electron wave function was lengthy process and problematic too.
- The basic approach of DFT is to develop exchange and correlation energy in terms of electron density

$$n(\mathbf{r}) = N \int \int \dots \int \psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N)^* \psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) d^3 r_1 d^3 r_2 \dots d^3 r_N \quad (7)$$

## Kohn-Sham equation

Kohn-Sham equation can be written as,

$$\left[ -\frac{1}{2} \nabla_i^2 + V_{\text{eff}}(\mathbf{r}_i) \right] \psi_i = \epsilon_i \psi_i \quad (8)$$

Where  $V_{\text{eff}}$  is Effective Kohn-Sham potential

## Local Density Approximation (LDA)

$$E_{xc}^{LDA}[n] = \int n(\mathbf{r}) \epsilon_{xc}[n(\mathbf{r})] d\mathbf{r}. \quad (9)$$

where  $\epsilon_{xc}[n(\mathbf{r})]$  is a exchange- correlation energy per particle of a uniform gas of interacting electrons of density  $n(r)$ .

## Computational Approach

- The mode of study will be computational which will be conducted using WEIN2K package. The WEIN2K package is a computer program which performs quantum mechanical calculations on periodic solids and it is written in Fortran.
- For some cases we will use Quantum Espresso(QE) whenever it required.

# Expected Outcomes

We expect to find out the band structure of  $\text{La}_2\text{MMnO}_6$  ( $\text{M}=\text{Co},\text{Ni}$ ) compound and some of the structural, electronic and magnetic properties of  $\text{La}_2\text{MMnO}_6$  ( $\text{M}=\text{Co},\text{Ni}$ ) compound.

# Time schedule

Work	Time Duration (in Months)					
	1-2	3-4	5-6	7-8	9-10	11-12
Literature Review						
Software Familiarization						
Data Enumeration & Calculation						
Data Analysis & Paper Publishing						
Thesis Writing & Documentation						



## References

- ① T. Tesfamichael, M. Roknuzzaman, C. Zhang, K. Ostrikov, A. Du, H. Wang, and L. Wang, *Scientific Reports* **9**, 718 (2019).
- ② E. L. Meyer, D. Mutukwa, N. Zingwe, and R. Taziwa, *Metals* **8**, 667 (2018).
- ③ M. Singh, K. Truong, S. Jandl, and P. Fournier, *Physical Review*.**79** (2009).
- ④ A. Kolchinskaya, P. Komissinskiy, M. Baghaie Yazdi, M. Vafaei Khanjani and D. Mikhailova, *et al. Physical Review*. **B 85**, 22 (2012).
- ⑤ Z.-Y. Wu, C.B. Ma, X.G. Tang, R. Li and Q.X. Liu, *et al. Nanoscale Research Letters*. **8**, 207 (2013).
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- ⑦ P. Kumar, N. K. Singh, G. Gupta, and P. Singh, *RSC Advanced* ,**6** (2016).
- ⑧ E. L. Meyer, D. Mutukwa, N. Zingwe, and R. Taziwa, *Metals* **8**, 667 (2018).

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